

# 極低濃度ラドン測定システムの開発

- 田阪茂樹,松原正也(岐阜大・総合情報メディアセンター)  
竹内康雄(東京大学・宇宙線研/神戸大・理)  
山澤弘実,森泉純(名古屋大・工)

共同利用研究経費 旅費30万円

神岡⇔岐阜・名古屋

柏⇔岐阜・名古屋

# 研究目的

背景:

SK実験の太陽ニュートリノ事象の主なBG  
純水中 $^{222}\text{Rn}$ 壊変生成物 $^{214}\text{Bi}$ の $\beta$ 線

「純水中ラドン起源BG解明と低減」

ニュートリノ事象の解析閾値を下げる

技術的課題:

純水中極低ラドン脱気

脱気ガス中ラドン吸着・脱離

高感度ラドン濃度測定

純水中極低濃度ラドン測定システム開発

→ $0.1\text{mBq/m}^3$

# Methods

## New Radon Detection System in Water :1)-5)

### 1) **Degassing Radon** in Purified Water:

New Air-Water Mixer:

Degassing Coefficient depend on Air/Water Flow Rate

### 2) **Dehumidified** Water Vapor:

Dehumidifier Efficiency by 4 Methods

### 3) **Absorption Radon** by Activated Charcoal:

Coefficient on Cooling Temperature:  $-90^{\circ}\text{C}$

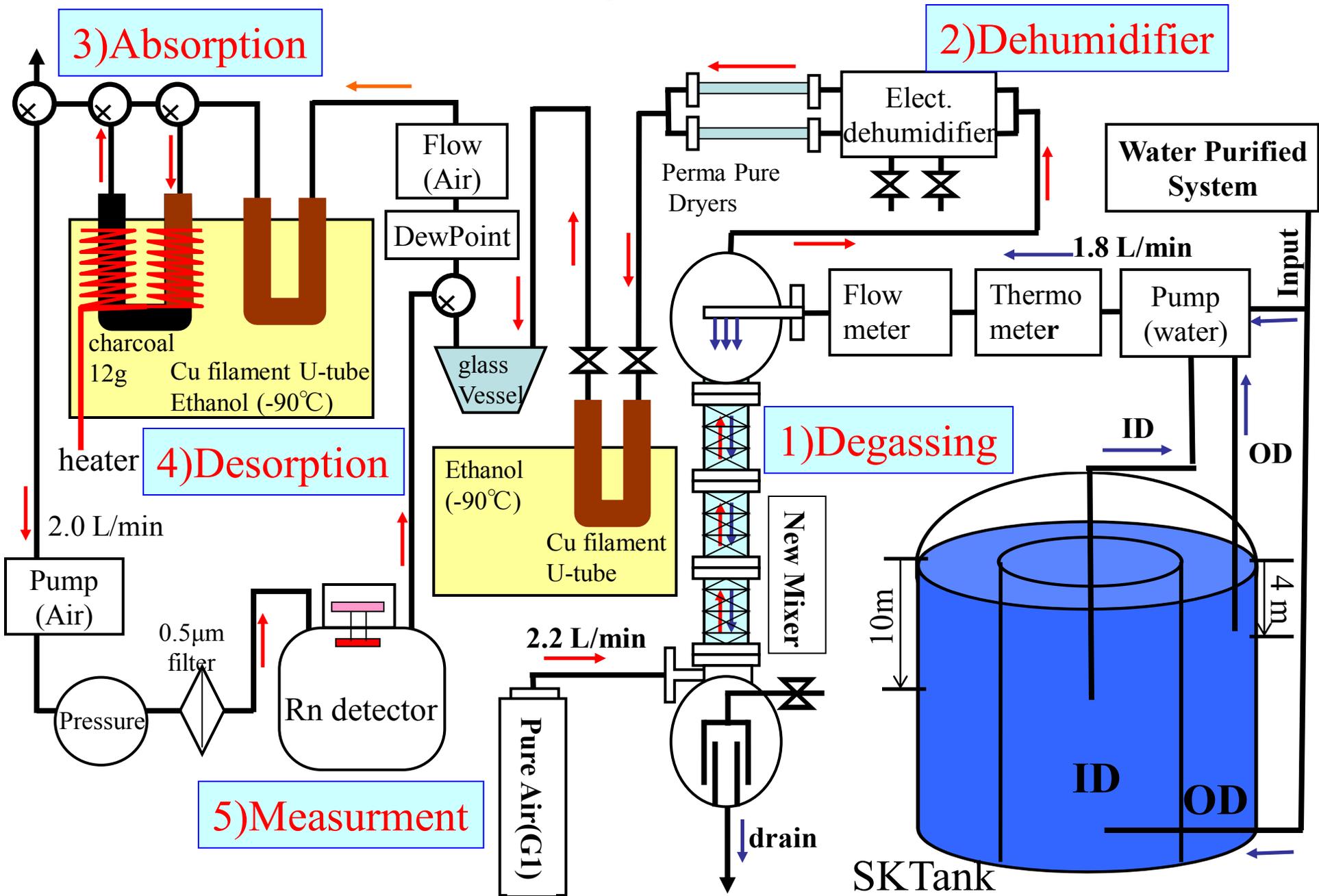
### 4) **Desorption Radon** by Activated Charcoal:

Coefficient on Heating Temperature :  $+250^{\circ}\text{C}$

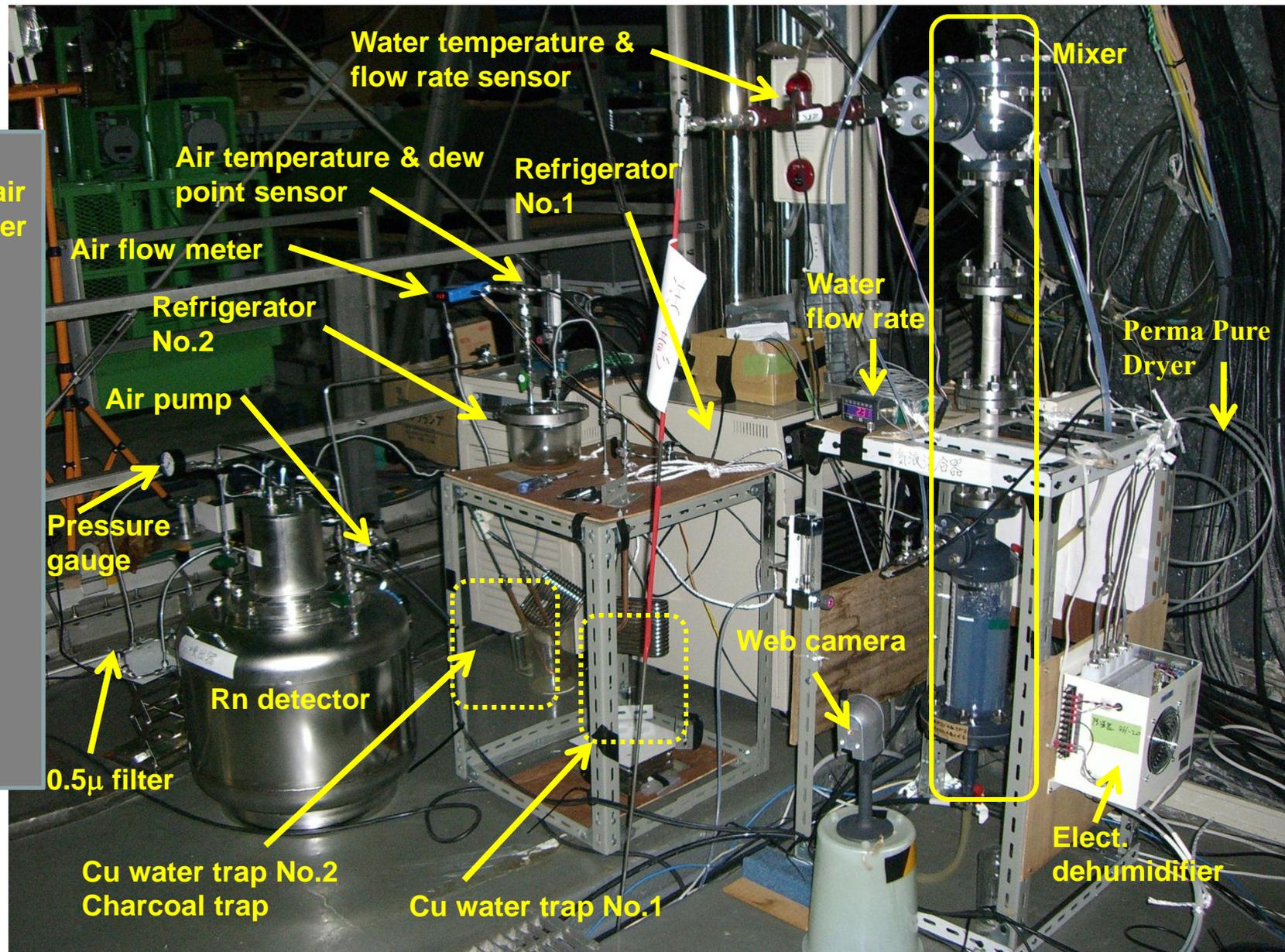
### 5) **Measurement Radon**

High Sensitivity Radon Detector with Electrostatic Collection of  $^{222}\text{Rn}$  Daughters( $^{214}\text{Po}$ ): Detection Limits  $5(\text{mBq}/\text{m}^3)$

# Radon Measurement System in SK Purified Water

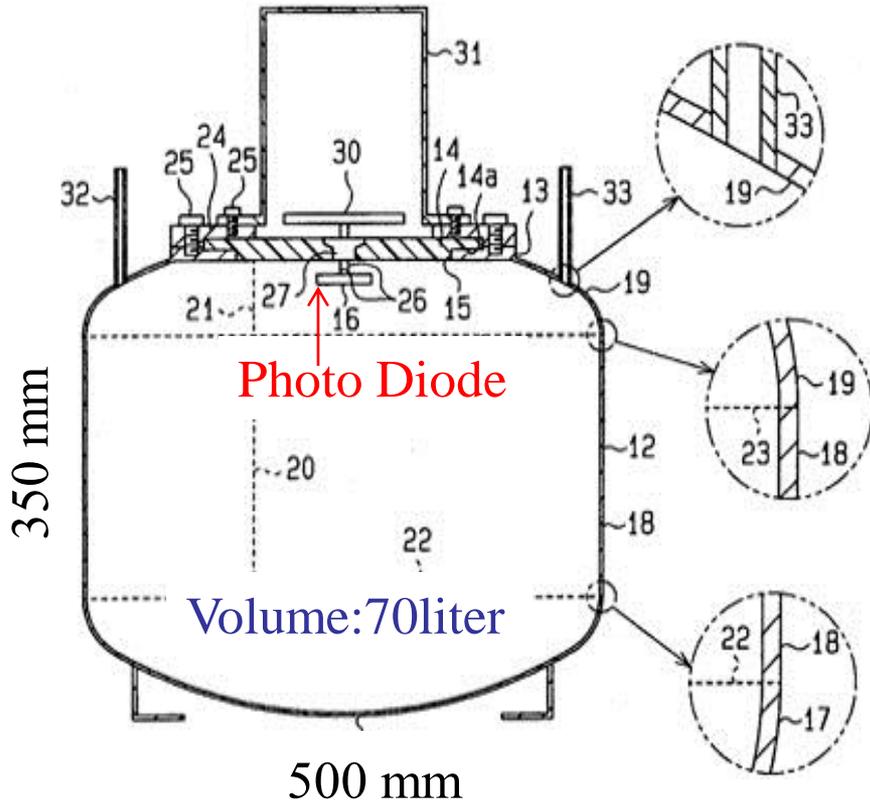


# Radon measurement system



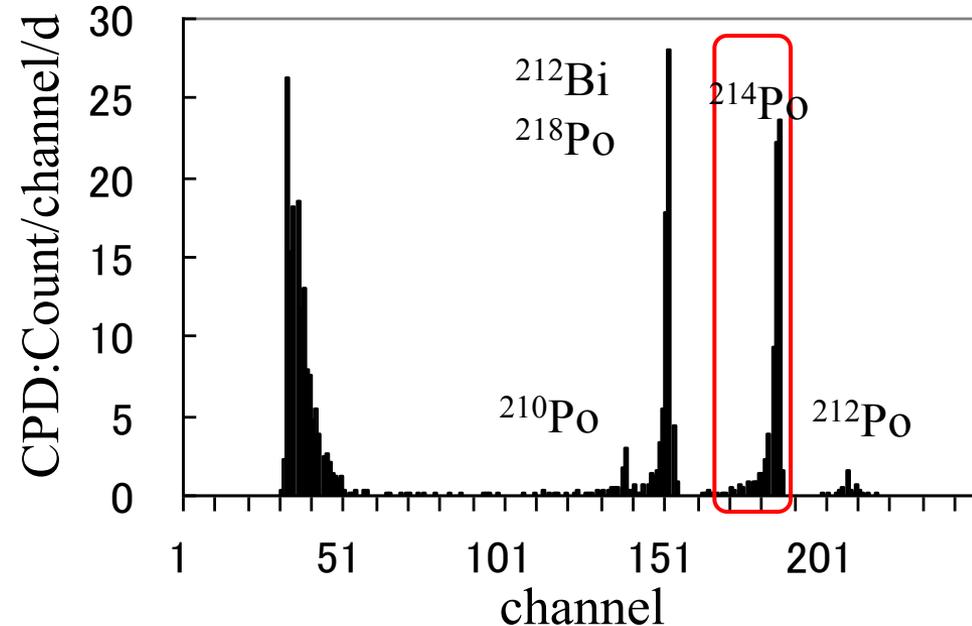
# 5) Measurement Radon

## Electrostatic collection Radon Detector



- Sensor : PIN Photo-Diode  
α-rays energy separation
- Electro-polished stainless steel
- Electrostatic Voltage -2kV
- Detection Limits 6(mBq/m<sup>3</sup>)

## Typical Pulse high distribution 2009/11/30-2009/12/11



## Detector Calibration by Source

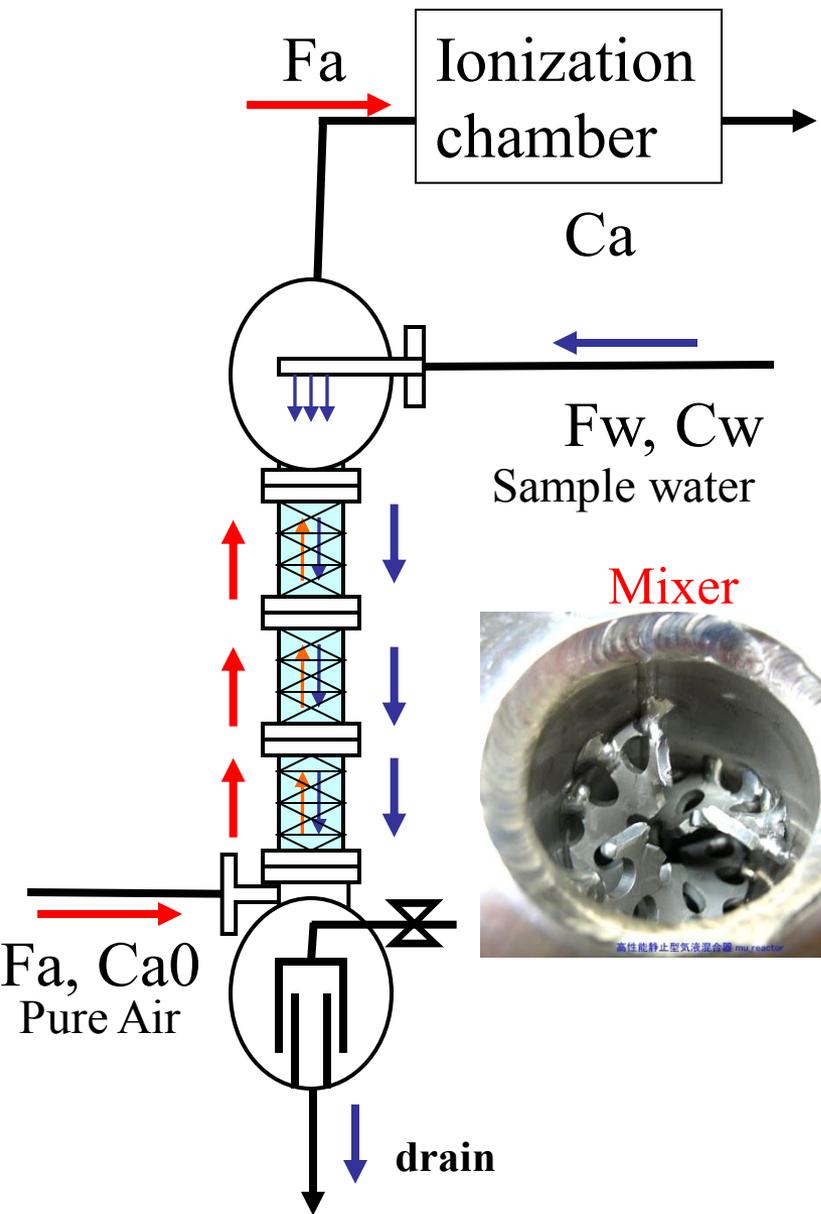
<sup>226</sup>Ra(PYLON 78.3Bq)

CF=1.93 ± 0.01 [CPD/(mBq/m<sup>3</sup>)]

AH: Absolute Humidity=0.011(g/m<sup>3</sup>)

CF=2.03-1.08\*SQRT(AH), AH<1.6g/m<sup>3</sup>

# 1) Degassing Radon in Purified Water



$$C_w = C_a \left( \frac{F_a}{F_w} + \alpha \right), \quad C_{a0} \gg C_w$$

F<sub>w</sub>: Water Flow = 1.83 L/min

F<sub>a</sub>: Air Flow = 2.32 L/min

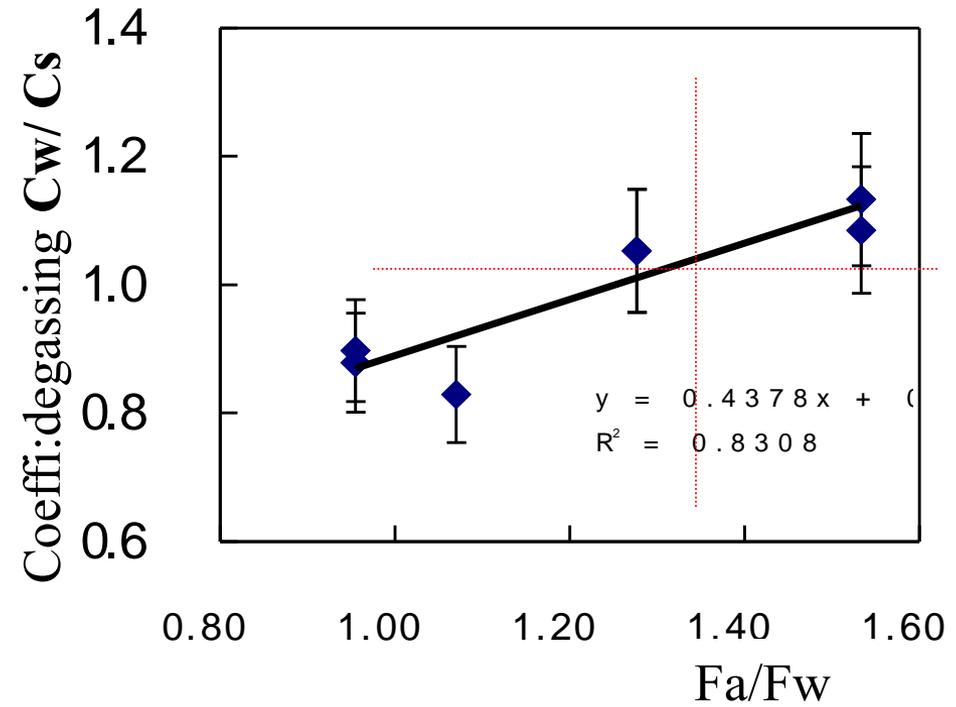
α: Solubility = 0.24

C<sub>a</sub>: Rn Concent. in Gas Phase

C<sub>w</sub>: Rn Concent. in Liquid Phase

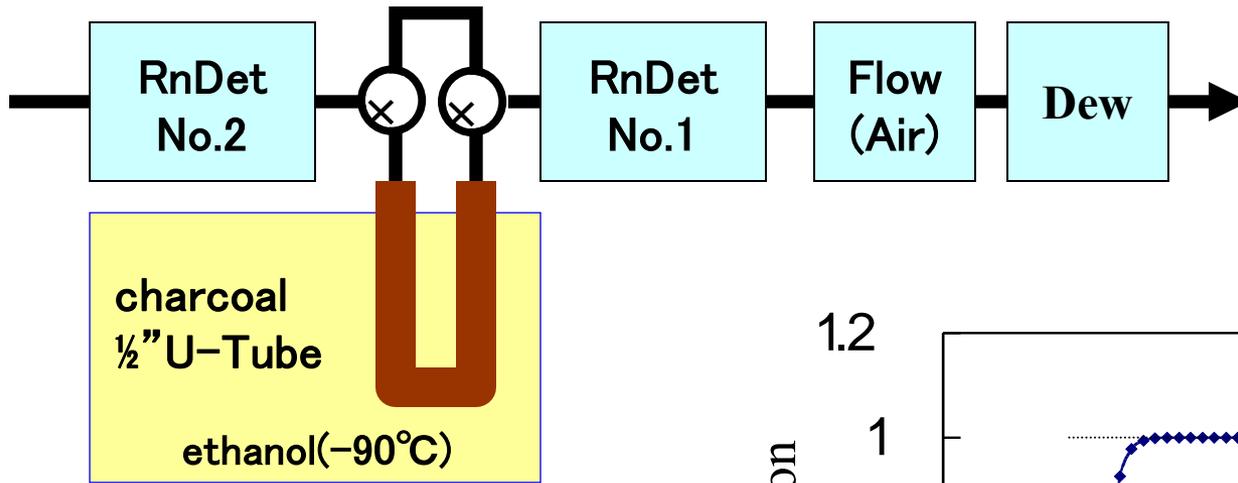
C<sub>s</sub>: Water Rn Concent. (10 Bq/L)

by Liquid Scintillation Method from ISO



# 3) Absorption Radon by Charcoal

Absorption Coefficient with Cooling Temp  $-90^{\circ}\text{C}$

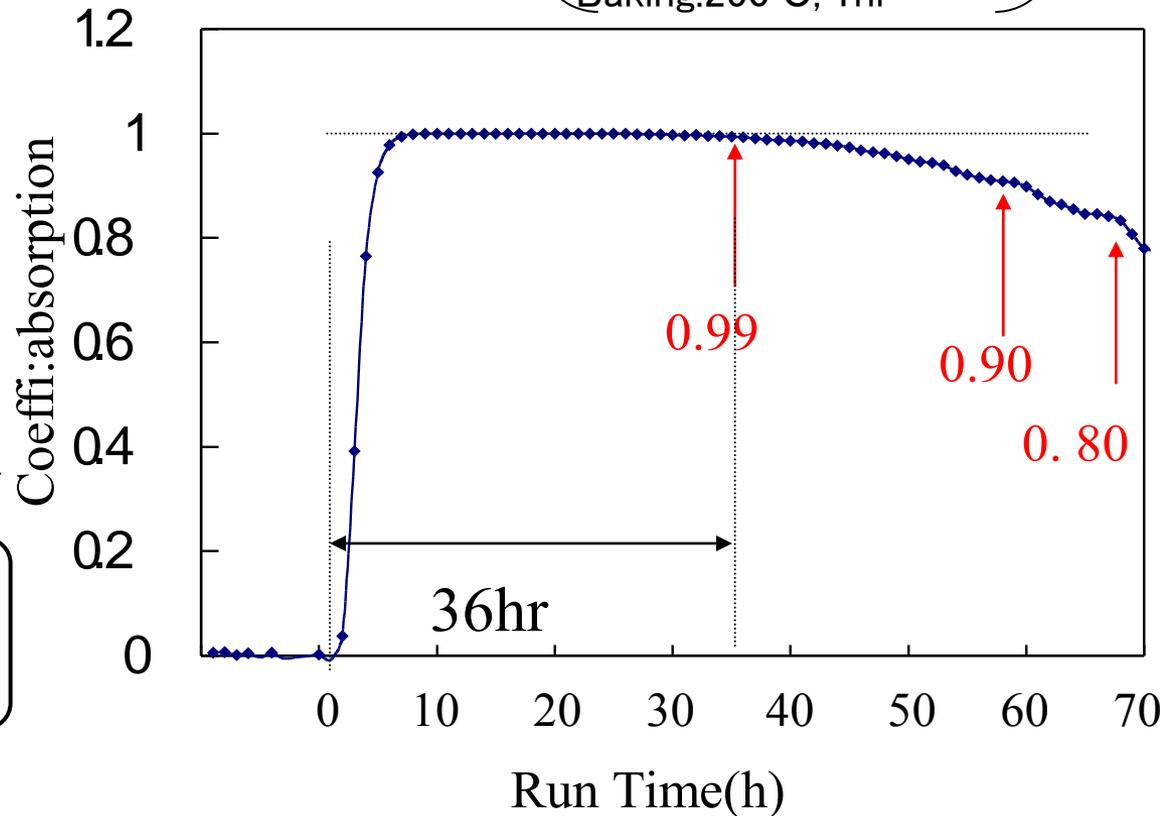


Exp conditions  
 Sample air: mine air  
 ( $100 \text{ Bq/m}^3$ )  
 DewPoint:  $-55^{\circ}\text{C}$   
 Fair: 2 L/min  
 charcoal: 24 g  
 Baking:  $250^{\circ}\text{C}$ , 1hr

charcoal:  
 Mitsubishi Chemicals: G4-5

$$\text{Coeffi:absor} = 1 - \frac{\text{CPH}_1}{\text{CPH}_2 \cdot D}$$

$\left[ \begin{array}{l} \text{CPH}_1 : \text{RnDetNo1:}^{214}\text{PoCPH} \\ \text{CPH}_2 : \text{RnDetNo2:}^{214}\text{PoCPH} \\ D : \text{Difference of Two Rn Det} \end{array} \right]$



# Radon Measurement, OD, Input and ID Water

RUN	Sample	Time(hr)	Va (m <sup>3</sup> )	Vw (m <sup>3</sup> )	<sup>214</sup> Po (CPD)	Ca (mBq/m <sup>3</sup> )	Cw (mBq/m <sup>3</sup> )
*1	BLANK	5	0.66	-	40.7	* 21.59	-
*2	BLANK	9	1.16	-	32.0	* 16.70	-
3	OD	5	0.67	0.55	105.4	55.32	7.08
4	OD	7	0.94	0.77	125.0	65.38	6.53
5	OD	9	1.21	1.00	169.0	88.20	7.34
6	Input	12	1.78	1.31	60.0	32.02	1.06
7	Input	27	3.92	2.95	87.0	45.64	1.03
* 8	ID	27	3.76	2.94	80.0	42.08	0.82

$$F = F_{\text{mixer}} \times F_{\text{absorption}} \times F_{\text{desorption}} \times F_{\text{decay}}$$

$$Ca' = CPD / CF$$

$$Va / Vw = Fa / Fw = 1.21 \sim 1.36$$

$$\text{Enrichment: } Va / Vd = 10 \sim 56$$

$$Vd = 0.07 (\text{m}^3)$$

$$* C_{BG} = 19 \pm 3 (\text{mBq/m}^3)$$

$$\alpha = 0.29 (T_w = 16^\circ\text{C})$$

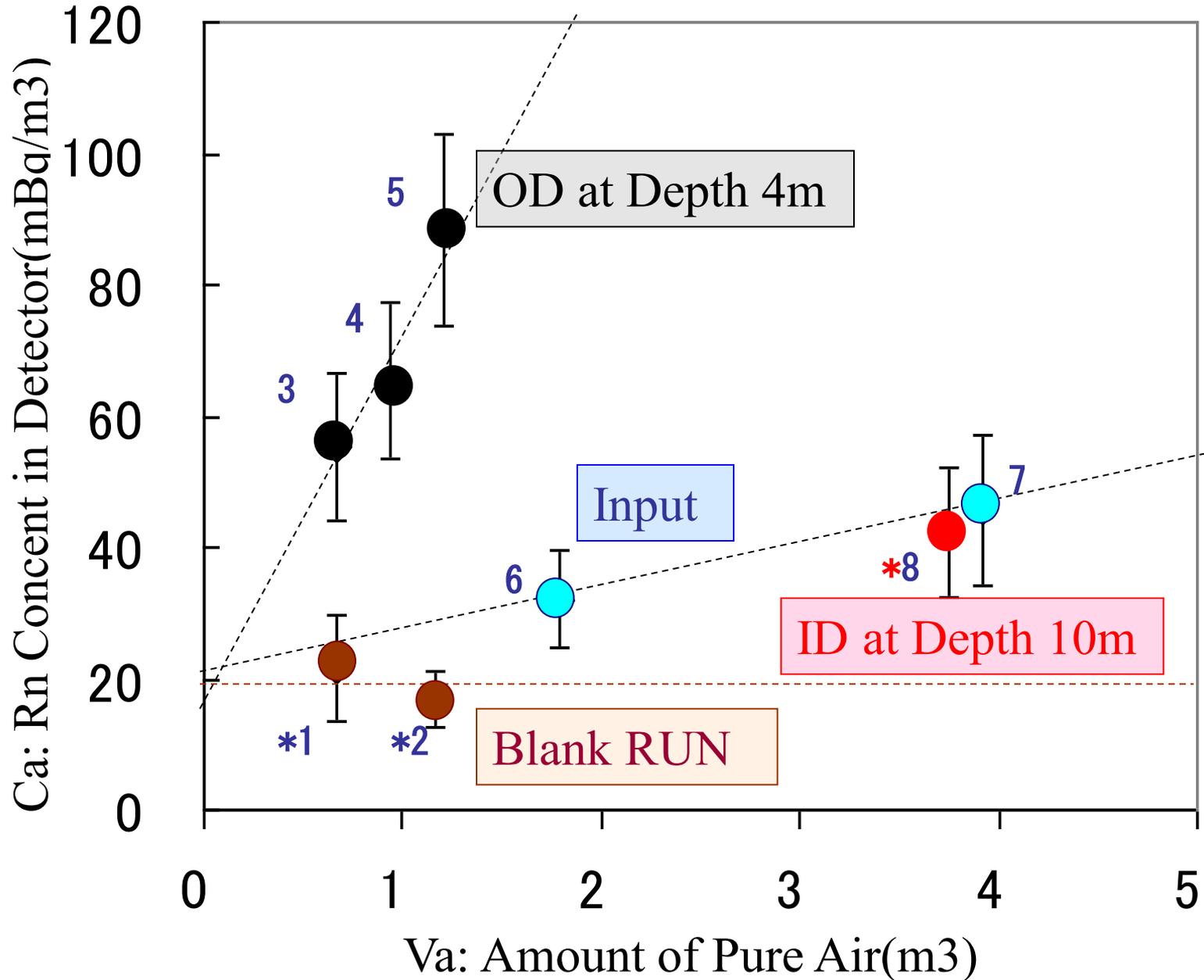
$$Cw = Ca (Va / Vw + \alpha)$$

$$Ca = (Vd / Va) (Ca' - C_{BG}) / F$$

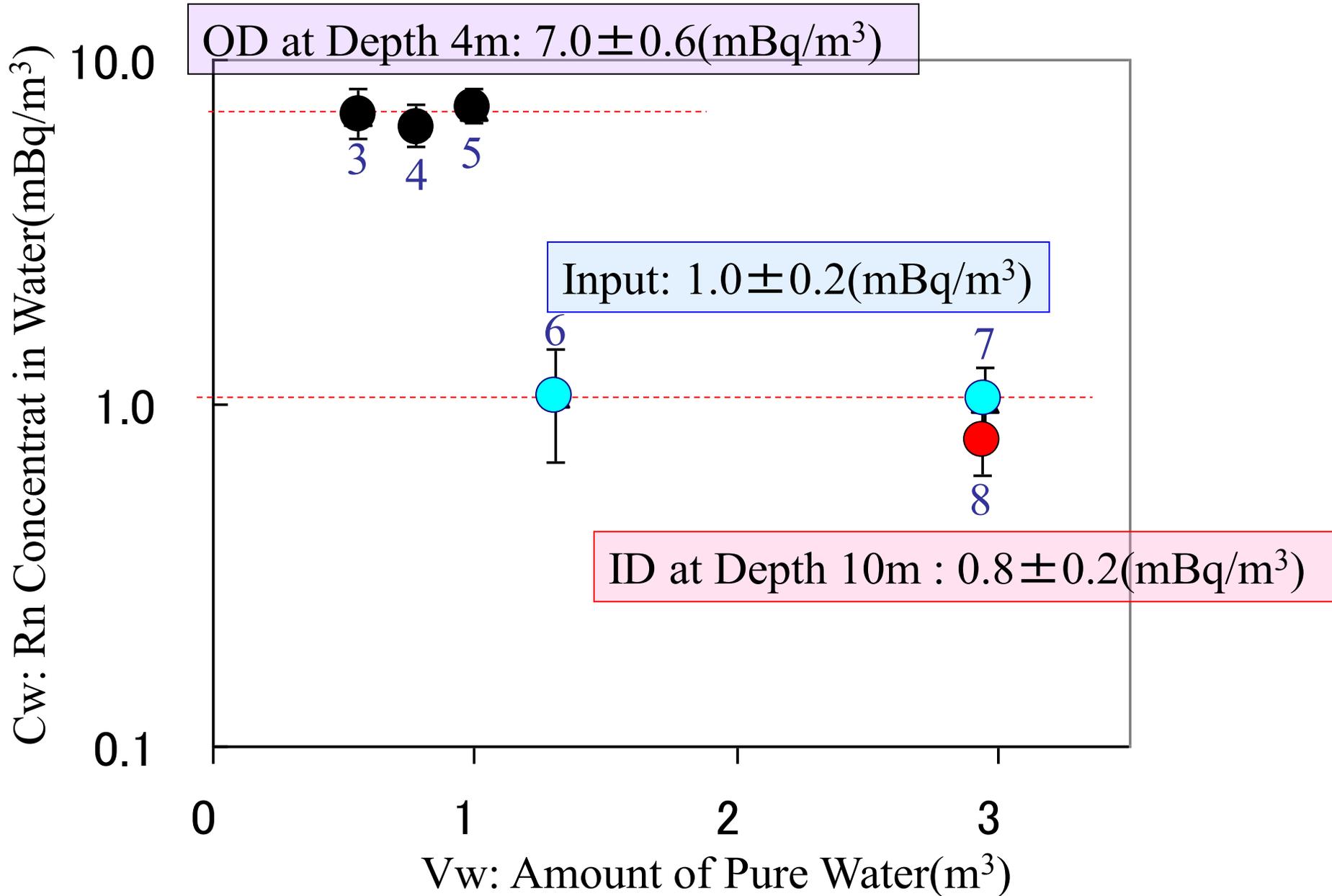
\*without Flushed Pure Water

\*preliminary

# Radon Concentration in SK Pure Water



# Radon Concentration in SK Pure Water



# Summary

## Development of Low Level Radon Concentration Measurement System in SK Pure Water, by using Air/Water New Mixer, and Absorption/Desorption of Activated Charcoal

### This Results:

- **OD Water at Depth 4m:  $7.0 \pm 0.6$  (mBq/m<sup>3</sup>)**
- **Input Water:  $1.0 \pm 0.2$  (mBq/m<sup>3</sup>)**
- **ID water at Depth 10m:  $0.8 \pm 0.2$  (mBq/m<sup>3</sup>)**

Phy.Lett.B452(1999)418-424, by Y.Takeuchi et al.

- Return Water  $<5.0$  (mBq/m<sup>3</sup>)
- Input Water  $<3.2$  (mBq/m<sup>3</sup>)
- ID at Tank Bottom  $3-5$  (mBq/m<sup>3</sup>)
- ID at Tank Central  $<1.4$  (mBq/m<sup>3</sup>)

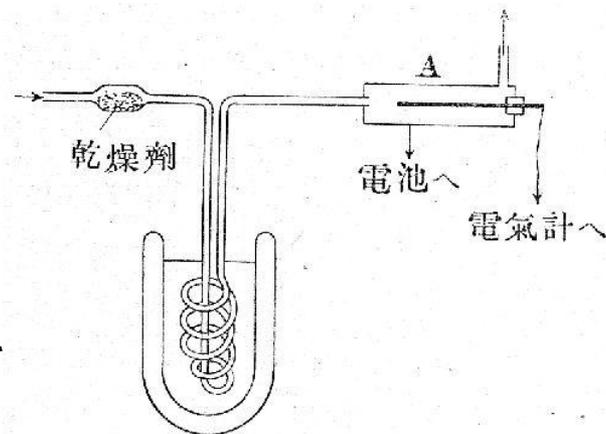
# 今後の改良

- 気液混合器(排水・気密性)
- 配管系 swagelok → VCR
- 活性炭(エタノール-90°C)  
→ 蛇管(液体窒素)

常温では固体の器壁はRnを吸収しないが低温では非常に強くそれに凝固する。…冷凍器が液体空気の温度(-180°C)にある時エマナチオンは全部蛇管内に凝固する。冷凍器の温度を再び上げると電離箱内に電流が現われるのを確かめることが出来る。これはRnに対しては-152°Cで急に起る。

マリーキュリー「放射能」176頁

9章放射性気体 § 46. 液体及び固体による  
吸収、溶解度、低温に於ける凝固



第 70 圖